2,4-D

With continued satisfactory operation of the 2,4-D Unit, technical activity in this area decreased. The quantity of each liquor being recycled to the reactor was increased by 20% to 600 gallons per batch, with continued beneficial results. Upon completion of the analysis of some samples now in the Lab, results will be summarized and this study will be brought to a close.

Installation of the recorder, measuring sodium sulfate flow to the 606 was completed, and it is generally operating satisfactorily. Data on the quantity being used to wash the batches now being taken will be used as a pasis for control of the sulfate flow.

An evaluation of the heat transfer rate in the present 2,4-D, reactor indicates that our rate of heat transfer has increased over that was experienced in the dimpled jacket vessel. Some additional work is being done to determine the maximum heat transfer coefficient that might be expected, and how it can be achieved.

Two miscellaneous studies in the 2,4-D area were completed during the month:

- 1. A long-term evaluation of the use of plastic-lined values to replace glass-lined values, which showed the plastic-lined values to be unsuitable at our service conditions. Plashon handling moltin 2,4-D at 150°C failed due to rupture of the lining, and Kyner at 105°C softened sufficiently that the lining at the usir was deformed, resulting in a leak.
- 2. An economic evaluation of the use of ECl to saidify 2,4-D in lieu of H2SO, showed this to be a highly unaconomic procedure. At the present 2,4-D volume, the use of HCl would increase around 2,4-D manufacturing costs by \$36,000.

HCA/DOP/HCL

Operation of the DCP separator was hampered by an imbility of the liquid to return continuously to the chlorinators, due to an imadequate liquid scal. Late in the month, the seal height was increased, but frequent plugging of the l' line still resulted in unsatisfactory operation. Additional corrective piping changes are currently being made.

Muriatic acid produced during March was of good quality. In early April, however, for some as yet unexplained reason, the phenol content in the acid rose considerably above specs. A study currently underway indicates that the Emount of chlorophenols discharged from the DCP Unit can vary markedly with chlorination conditions. Effort is now being given to determine how this variation and the unsatisfactory operation of the DCP separator relate to the current high phenol concentration in the Muriatic acid.

A study of the isomer flictuithation during a DCP chlorination was made. The isomer distribution when plotted against specific gravity gave a family of curves almost identical to curves developed in 1963 by Mr. R. F. Lindemann.

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One of the Toffon dip pipes installed in MA service last Fall failed due to a defect in manufacture. It was replaced by the vendor at no cost, and the new unit has been installed in the observator.

TOP

Tests to determine the effectiveness of p-dioxin removal indicated that the most active 2,3,7,8 isomer is the most easily absorbed of the impurities in the stream. A quantitative ensure on the effectiveness of carbon is not you evailable due to uncertainty over what part a filtration step in the test procedure was playing in the p-dioxin removal. Additional carbon tests will be run this month. Filtration tests will also be resumed, using a Public cartridge-type filter.

A preliminary evaluation was eade of piping changes necessary to convert Tank #131 for use as an anisole still. Further work in this area is smalting review of the proposed changes.

INCMEDIES

A number of empleifiers were evaluated in an effort to obtain comparable empleification in 2% Decembe at the lower same ratio (.333%/% acid) to what we are now getting. Several acceptable empleifiers were found. Major work on the December was directed toward proparing a December "wax" bar. A number of formulations were tried, and samples of several promising ones were sent to Cleveland for evaluation. We are available word on how the samples appear before proceeding further on this project.

EMPALISION (APPROPRIATION NO. 6739)

Construction work on the main process changes was started in March, with the main areas of activity being relocation of the ${\rm Cl}_2$ vaporizer, erection of the steel for the FCP and Flaker Building expansions, and work on the FCP transfer lines. Progress in the Marchouse and tank form areas placed drastically toward the end of the month. Adverse weather accounted for part of the delay, but a more significant reason for the slow-down use the small number of men available for assignment to these areas of work.

The rajor area of work in the engineering phase of the project was directed toward modification of the project schedule to permit operation of the "T" Unit into July. Detailed engineering continued on piping, electrical, etc.

MISCELLANEOUS

The proposed formulation for Singlechot (Crop Rider 322) was checked and found to be OK. The first samples were prepared in the Lab and subsequently, about 180 gallons were prepared in the Plant and packaged in one-gallon cans for use in sample shipments.

Mixed mono, di, and trichloroacetic acids were prepared in the Lab, with no particular problem. The major change between this chlorimation and a standard MCA run was that the pot temperature had to be significantly of in the late stages of the chlorimation to keep the chlorine efficiency at an acceptable level. A manufacturing cost estimate for the preparation of the mixed chloroacetic scias was prepared based on the results of the lab chlorimation.

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DS 00001047

A visit was made to Cangill labs to pick up some Termil for shipment, and also wake a cursory evaluation of how the gueduction openation was going.

A pollution survey listing all plant offluents was completed, and it has now been forwarded to Central Engineering for inclusion in the Corporate report to the Lanufacturing Chemists Association. Two representatives also attended on Air Pollution Conference hold in Philadelphia.

One personnel change in the Technical Group occurred early in the month with the reassignment of J. J. Imanual to the Saul-Works. He was replaced by Mr. D. R. Gebran who joined the Plant on Month 5th.

No appropriations were closed during the renth.

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FES/mc

4/12/67

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OFTRATUIG DATA - MARCH, 1957

2,4-0	"D" SIDE	"T" SID	TO	TALS
Average DOP Conversion, \$\partial Average Cycle Time, Hours Average Cooking Time, Hours Average Cooking Temperature, °C Usage #/# Product, DOP/:DA Average Product Assay, \$\partial P	95.4 9.0 1.9 102.5 .853/.604 99.4	No Producti This Side	lon 10 .85	9.0 9.0 1.9 2.5 3/.604
MCA/DCP				
Number of Batches Average Batch Size, Lbs. Average Reaction Time, Hours Average/Maximum Reaction Temp. °C Average Emit Gas Temperature, °C Usage #/# Product, Chlorine Usage #/# Froduct, Acetic or Phenol Product Assay, \$ MCA -) Acetic -) Anhydride -)	54 8,303 7.1 106/120 -18 .457 .595 None Assayed	2,6-D0	13. 80/ 20 .49 .57 P - 94. P - 1.	99 5 4 5
EGT .				
Average Phanol Content, pyn Average Sulfate Content, ppm	63)Levels were reduced by addition 33)of purchased sold.			
2,4,5-T				
Average TCP Conversion, \$ Average Cycle Time, Hours Average Cooking Time, Kours Average Cooking Temperature, °C Usage #/# TCP/KCA Average Product Assay, \$	73.5 Not deter 4.8 104 .902/.573 Home Asso			esete :
TCP			TCP ASSA	Y
Number of Betches Average Eatch Sine, Lbs. Average Reaction/Digestion Time, Frs. Average/Huminum Autoclave Temp. °C Huminum Temp.in Anisole Still, °C Usage #/# Product, TyCB Kethanol Caustic(Lig./Solid)	54 2,458 1.9/5.0 167/175 105 1.047 .502 .415/.337	2,4,5-TCP -) 2,4-DCP -) None 2,6-DCP -)Assayed Amisole -)		
<u>ESTETS</u>	BUTYL-D	BUTYL-T	2-FA-D	2-EH-T
Number Eatches Average Batch Size, Lbs. Average Cycle Time, Hours Average Reaction Temperature Average Free Acid, % D\$ 00000 Average Color	49 7,251 25.3 4 9 0.7 1.8	15 5,608 22.5 139 0.0(ASTM)	11 7,265 20.7 156 1.3	11 6,231 24.7 154 2.0
FGS/nc 4/12/67				

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