

Right to Know Hazardous Substance Fact Sheet

Common Name: ANILINE

Synonyms: Aminobenzene; Phenylamine

Chemical Name: Benzenamine

Date: June 1998 Revision: January 2008

Description and Use

Aniline is a colorless to brown, oily liquid. It is used in making dyes, herbicides, drugs, photographic chemicals, rubber accelerators, inks, perfumes, resins and varnishes.

▶ ODOR THRESHOLD=0.58 to 10 ppm

 Odor thresholds vary greatly. Do not rely on odor alone to determine potentially hazardous exposures.

Reasons for Citation

- ► Aniline is on the Right to Know Hazardous Substance List because it is cited by OSHA, ACGIH, DOT, NIOSH, DEP, IARC, IRIS, NFPA and EPA.
- ► This chemical is on the Special Health Hazard Substance List.

SEE GLOSSARY ON PAGE 5.

FIRST AID

Eye Contact

▶ Immediately flush with large amounts of cool water for at least 30 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while flushing. Seek medical attention immediately.

Skin Contact

Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of soap and water. Seek medical attention immediately.

Inhalation

- ▶ Remove the person from exposure.
- Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- ▶ Transfer promptly to a medical facility.

EMERGENCY NUMBERS

Poison Control: 1-800-222-1222 CHEMTREC: 1-800-424-9300 NJDEP Hotline: 1-877-927-6337

National Response Center: 1-800-424-8802

CAS Number: 62-53-3
RTK Substance Number: 0135
DOT Number: UN 1547

EMERGENCY RESPONDERS >>>> SEE BACK PAGE

Hazard Summary Hazard Rating NJDOH NFPA HEALTH - 2 FLAMMABILITY - 2 REACTIVITY - 0

CARCINOGEN COMBUSTIBLE

POISONOUS GASES ARE PRODUCED IN FIRE

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- ► Aniline can affect you when inhaled and by rapidly passing through the skin.
- ► Aniline should be handled as a CARCINOGEN--WITH EXTREME CAUTION.
- ▶ Aniline can cause skin irritation and rash.
- ▶ Aniline can irritate the eyes with possible eye damage.
- ▶ Inhaling Aniline can irritate the nose and throat.
- ► High levels of this substance can reduce the blood's ability to transport Oxygen, causing headache, fatigue, dizziness, and a blue color to the skin and lips (methemoglobinemia). Exposure to very high levels can cause trouble breathing, collapse and even death.
- ► Aniline may damage the liver, cause anemia and affect the nervous system.

Workplace Exposure Limits

OSHA: The legal airborne permissible exposure limit (PEL) is **5 ppm** averaged over an 8-hour workshift.

NIOSH: Recommends that exposure to occupational carcinogens be limited to the lowest feasible concentration.

ACGIH: The threshold limit value (TLV) is **2 ppm** averaged over an 8-hour workshift.

- ▶ Aniline may be a CARCINOGEN in humans. There may be no safe level of exposure to a carcinogen, so all contact should be reduced to the lowest possible level.
- ▶ The above exposure limits are for air levels only. When skin contact also occurs, you may be overexposed, even though air levels are less than the limits listed above.

ANILINE Page 2 of 6

Determining Your Exposure

- ▶ Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- ► For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility's RTK Central File or Hazard Communication Standard file.
- ➤ You have a right to this information under the New Jersey Worker and Community Right to Know Act, the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- ► The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) requires private employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

Health Hazard Information

Acute Health Effects

The following acute (short-term) health effects may occur immediately or shortly after exposure to **Aniline**:

- ▶ Aniline can cause skin irritation and rash.
- ▶ Aniline can irritate the eyes with possible eye damage.
- ▶ Inhaling **Aniline** can irritate the nose and throat causing coughing and wheezing.
- ▶ High levels of this substance can reduce the blood's ability to transport *Oxygen*, causing headache, fatigue, dizziness, and a blue color to the skin and lips (*methemoglobinemia*). Exposure to very high levels can cause trouble breathing, collapse and even death.

Chronic Health Effects

The following chronic (long-term) health effects can occur at some time after exposure to **Aniline** and can last for months or years:

Cancer Hazard

- ► Aniline may be a CARCINOGEN in humans since it has been shown to cause spleen cancer in animals.
- ► Many scientists believe there is no safe level of exposure to a carcinogen.

Reproductive Hazard

- ▶ Aniline may damage the developing fetus.
- ► There is limited evidence that women employed in the **Aniline** dye industry have a higher incidence of reproductive cycle disorders and abortion.

Other Effects

- ▶ High exposure may damage the liver.
- ► Aniline may cause anemia.
- ► Exposure may affect the nervous system causing irritability, fatigue, drowsiness, convulsions (fits), disturbance of speech, upset stomach, muscle pain and unconsciousness.

Medical

Medical Testing

For frequent or potentially high exposure (half the TLV or greater, or significant skin contact) the following are recommended before beginning work and at regular times after that:

- ▶ Liver function tests
- ► Complete blood count

If symptoms develop or overexposure is suspected, the following are recommended:

- ▶ Blood methemoglobin level
- ▶ Exam of the nervous system

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are <u>not</u> a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

Mixed Exposures

▶ Because more than light alcohol consumption can cause liver damage, drinking alcohol may increase the liver damage caused by Aniline. ANILINE Page 3 of 6

Workplace Controls and Practices

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at www.cdc.gov/niosh/topics/ctrlbanding/.

The following work practices are also recommended:

- ▶ Label process containers.
- ▶ Provide employees with hazard information and training.
- Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- ▶ Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material
- ▶ Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- ▶ Do not take contaminated clothing home.
- ▶ Get special training to wash contaminated clothing.
- ▶ Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

Personal Protective Equipment

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

Gloves and Clothing

- Avoid skin contact with Aniline. Wear personal protective equipment made from material which can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.
- ➤ Safety equipment manufacturers recommend Butyl, Silver Shield®/4H® and Viton for gloves and DuPont Tychem® CPF 2, CPF 4, BR, LV, SL, TK and Responder®, Kappler Zytron® 200; and Saint-Gobain ONESuit®TEC or equivalent as protective materials for clothing.
- ▶ All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

Eye Protection

- Wear indirect-vent, impact and splash resistant goggles when working with liquids.
- ► Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.
- ▶ Do not wear contact lenses when working with this substance.

Respiratory Protection

Improper use of respirators is dangerous. Respirators should only be used if the employer has a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- ▶ Where the potential exists for exposure over **2 ppm**, use a NIOSH approved full facepiece respirator with an organic vapor cartridge. Increased protection is obtained from full facepiece powered-air purifying respirators.
- ▶ Leave the area immediately if (1) while wearing a filter or cartridge respirator you can smell, taste, or otherwise detect **Aniline**, (2) while wearing particulate filters abnormal resistance to breathing is experienced, or (3) eye irritation occurs while wearing a full facepiece respirator. Check to make sure the respirator-to-face seal is still good. If it is, replace the filter or cartridge. If the seal is no longer good, you may need a new respirator.
- ► Consider all potential sources of exposure in your workplace. You may need a combination of filters, prefilters or cartridges to protect against different forms of a chemical (such as vapor and mist) or against a mixture of chemicals.
- ▶ Where the potential exists for exposure over **20 ppm**, use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus operated in a pressure-demand or other positive-pressure mode.
- ▶ Exposure to **100 ppm** is immediately dangerous to life and health. If the possibility of exposure above **100 ppm** exists, use a NIOSH approved self-contained breathing apparatus with a full facepiece operated in a pressure-demand or other positive-pressure mode equipped with an emergency escape air cylinder.

Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- ▶ Aniline is a COMBUSTIBLE LIQUID.
- ► Use dry chemical, CO₂, water spray or foam as extinguishing agents.
- ▶ POISONOUS GASES ARE PRODUCED IN FIRE, including Nitrogen Oxides.
- ▶ Use water spray to keep fire-exposed containers cool.
- ► Vapor is heavier than air and may travel a distance to cause a fire or explosion far from the source.

ANILINE Page 4 of 6

Spills and Emergencies

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If Aniline is spilled or leaked, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- ▶ Eliminate all ignition sources.
- ► Absorb liquids in vermiculite, dry sand, earth, or a similar material and deposit in sealed containers.
- ▶ Ventilate and wash area after clean-up is complete.
- ▶ It may be necessary to contain and dispose of **Aniline** as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

Handling and Storage

Prior to working with **Aniline** you should be trained on its proper handling and storage.

- ▶ Aniline reacts violently with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE) and STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC) and may cause fires and explosions.
- ▶ Aniline is not compatible with ACETIC ANHYDRIDE; CHLOROSULFONIC ACID; STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC); ALKALIES (such as METAL HYDROXIDES and METAL CARBONATES); STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE); and TOLUENE DIISOCYANATES.
- ► Store in tightly closed containers in a cool, well-ventilated area away from AIR, LIGHT and COPPER.
- Sources of ignition, such as smoking and open flames, are prohibited where **Aniline** is used, handled, or stored in a manner that could create a potential fire or explosion hazard.
- ► Store under *Methanol* to prevent polymerization.
- ▶ Aniline attacks RUBBER, PLASTIC, and COATINGS.

Occupational Health Information Resources

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

For more information, please contact:

New Jersey Department of Health Right to Know

PO Box 368

Trenton, NJ 08625-0368 Phone: 609-984-2202 Fax: 609-984-7407

E-mail: rtk@doh.state.nj.us

Web address: http://www.nj.gov/health/eoh/rtkweb

The Right to Know Hazardous Substance Fact Sheets are not intended to be copied and sold for commercial purposes.

ANILINE Page 5 of 6

GLOSSARY

ACGIH is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

Acute Exposure Guideline Levels (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

Boiling point is the temperature at which a substance can change its physical state from a liquid to a gas.

A **carcinogen** is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

CFR is the Code of Federal Regulations, which are the regulations of the United States government.

A combustible substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

DEP is the New Jersey Department of Environmental Protection.

DOT is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

EPA is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

ERG is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

Emergency Response Planning Guideline (ERPG) values are intended to provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

IARC is the International Agency for Research on Cancer, a scientific group.

Ionization Potential is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

IRIS is the Integrated Risk Information System database maintained by federal EPA. The database contains information on human health effects that may result from exposure to various chemicals in the environment.

LEL or **Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m³ means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

NFPA is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

NIOSH is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

NTP is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

OSHA is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

PEOSHA is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

Permeated is the movement of chemicals through protective materials.

PIH is a DOT designation for chemicals which are Poison Inhalation Hazards.

ppm means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

STEL is a Short Term Exposure Limit which is usually a 15-minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

UEL or **Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

Vapor Density is the ratio of the weight of a given volume of one gas to the weight of another (usually *Hydrogen*), at the same temperature and pressure.

The **vapor pressure** is a measure of how readily a liquid or a solid mixes with air at its surface. A higher vapor pressure indicates a higher concentration of the substance in air and therefore increases the likelihood of breathing it in.



Right to Know Hazardous Substance Fact Sheet

Emergency Responders **Quick Reference**

Common Name: ANILINE

Synonyms: Aminobenzene; Phenylamine

CAS No: 62-53-3

Molecular Formula: C₆H₅NH₂ RTK Substance No: 0135

Description: Colorless to brown, oily liquid

HAZARD DATA		
Hazard Rating	Firefighting	Reactivity
2 - Health	COMBUSTIBLE LIQUID	Aniline reacts violently with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE) and STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC) and may cause
2 - Fire	Use dry chemical, CO ₂ , water spray or foam	
0 - Reactivity	as extinguishing agents. POISONOUS GASES ARE PRODUCED IN	
DOT#: UN 1547	FIRE, including Nitrogen Oxides.	fires and explosions.
ERG Guide # : 153	Use water spray to keep fire-exposed containers cool.	Aniline is not compatible with ACETIC ANHYDRIDE; CHLOROSULFONIC ACID; STRONG ACIDS (such as
Hazard Class: 6.1 (Poisonous)	Vapor is heavier than air and may travel a distance to cause a fire or explosion far from the source.	HYDROCHLORIC, SULFURIC and NITRIC); ALKALIES (such as METAL HYDROXIDES and METAL CARBONATES); STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE); and TOLUENE DIISOCYANATES.

SPILL/LEAKS

Isolation Distance:

Small Spills: 60 meters (200 feet) Large Spills: 270 meters (900 feet)

Fire: 800 meters (1/2 mile)

Absorb liquids in vermiculite, dry sand, earth, or a similar material and deposit in sealed containers.

This substance is very toxic to aquatic organisms.

PHYSICAL PROPERTIES

Odor Threshold: 0.58 to 10 ppm Flash Point: 158°F (70°C)

LEL: 1.3% **UEL:** 11% Vapor Density: 3.2 (air = 1)

Vapor Pressure: 0.6 mm Hg at 68°F (20°C)

1 (water =1) **Specific Gravity:** Water Solubility: Slightly soluble **Boiling Point:** 363°F (184°C)

Ionization Potential: 7.7 eV **Molecular Weight:** 93.1

EXPOSURE LIMITS

OSHA: 5 ppm, 8-hr TWA

NIOSH: Lowest feasible concentration

ACGIH: 2 ppm, 8-hr TWA

IDLH LEVEL: 100 ppm

PROTECTIVE EQUIPMENT

Gloves: Butyl, Silver Shield®/4H® and Viton (>8-hr breakthrough)

Coveralls: DuPont Tychem® CPF 2, CPF 4, BR, LV, SL, TK and

Responder®; Kappler Zytron® 200; and Saint-Gobain ONESuit®TEC or equivalent (>8-hr breakthrough)

Respirator: >2 ppm - full facepiece APR with Organic vapor cartridge

>20 ppm - Supplied air

HEALTH EFFECTS

Eves: Irritation with possible eye damage

Skin: Irritation and skin rash

Inhalation: Nose and throat irritation with coughing

and wheezing

Headache, dizziness and blue color to the skin and lips (methemoglobinemia)

Fatigue, drowsiness, convulsions, disturbance of speech, upset stomach

and unconsciousness

FIRST AID AND DECONTAMINATION

Remove the person from exposure.

Flush eyes with large amounts of water for at least 30 minutes. Remove contact lenses if worn. Seek medical attention immediately.

Quickly remove contaminated clothing. Wash contaminated skin with large amounts of soap and water. Seek medical attention immediately.

Begin artificial respiration if breathing has stopped and CPR if necessary.

Transfer to a medical facility.